

CVD of CrO₂ Thin Films: Influence of the Deposition Parameters on their Structural and Magnetic Properties

A.F. Mota¹, A.J. Silvestre², P.M. Sousa¹, O. Conde^{1a}, M.A. Rosa³, M. Godinho³

¹Dep. Física/ICEMS, Faculdade de Ciências da Universidade de Lisboa, PT 1749-016 Lisboa

²Instituto Superior de Engenharia de Lisboa/ICEMS, PT 1950-062 Lisboa

¹Dep. Física/CFMC, Faculdade de Ciências da Universidade de Lisboa, PT 1749-016 Lisboa

^aoconde@fc.ul.pt

Keywords: Chromium dioxide (CrO₂), Atmospheric pressure CVD, Spintronics

Abstract. This work reports on the synthesis of CrO₂ thin films by atmospheric pressure CVD using chromium trioxide (CrO₃) and oxygen. Highly oriented (100) CrO₂ films containing highly oriented (0001) Cr₂O₃ were grown onto Al₂O₃(0001) substrates. Films display a sharp magnetic transition at 375 K and a saturation magnetization of 1.92 μ_B/f.u., close to the bulk value of 2 μ_B/f.u. for the CrO₂.

Introduction

Magnetic materials exhibiting a high degree of spin polarization are being actively investigated for their potential use in spintronic devices. Among them, chromium dioxide (CrO₂) is very attractive because it is a half-metal fully spin polarized at the Fermi level with Curie temperature above room temperature (T_c=393 K) [1-4]. CrO₂ has the rutile structure with a tetragonal unit cell (a=b=0.4421 nm, c=0.2916 nm) consisting of two formula units, the Cr ions being in the Cr⁴⁺ oxidation state, and a magnetic moment of 2 μ_B/f.u. [5].

The ability to produce CrO₂ films is crucial both for applications and fundamental studies. Nevertheless, it remains a challenging task to grow thin films of CrO₂ due to its metastability. CrO₂ easily decomposes into the stable Cr₂O₃ insulating antiferromagnetic oxide phase. Up to now, chemical vapour deposition (CVD) based on the methodology described by Ishibashi et al. [6] seems to be the most successful technique to produce such films. High quality CrO₂ films have been deposited on single-crystal rutile (TiO₂) [6-10], which has the same crystal structure and nearly the same lattice parameters as CrO₂. Sapphire has also been used as substrate material [11], however the films grown on this substrate have been shown to be contaminated with Cr₂O₃.

In this work, we focus on the synthesis and properties of CrO₂ thin films on Al₂O₃(0001) substrates by atmospheric pressure CVD. The study of the deposition process as a function of the deposition time allowed us to give an explanation for the occurrence of a Cr₂O₃ layer at the CrO₂ film/substrate interface on the basis of structural effects.

Experimental

The CrO₂ films were deposited onto Al₂O₃(0001) substrates by atmospheric pressure CVD following the method of Ishibashi et al. [6]. However, instead of the two-zone furnace used by these authors, we used a quartz tube placed inside a single-zone furnace and an independent control of the substrate temperature, similar to the system described in ref. [12]. Chromium trioxide (CrO₃) powder (purity 99.9%) was used as precursor and loaded into a stainless steel boat aligned with the substrate holder, 6 cm apart from it. Prior to their insertion into the reactor, the substrates were ultrasonically cleaned in organic solvents, rinsed in distilled water and dried with a N₂ flux. Oxygen (purity 99.999%) was used as carrier gas.

Films were grown for the experimental parameters presented in table 1. In order to avoid the deposition of any kind of impurities during the initial stages of the deposition process, the substrate

was always heated up to the deposition temperature before the melting of the CrO_3 precursor occurred ($T_m=196\text{ }^\circ\text{C}$). Best results were obtained for substrate temperature $T_s = 390\text{ }^\circ\text{C}$, precursor temperature $T_p = 260\text{ }^\circ\text{C}$ and oxygen flow rate $\phi_{\text{O}_2} = 280\text{ sccm}$. In this paper we will concentrate on the analysis of films grown with these experimental parameters for deposition times ranging from 1 to 6 h.

Table 1 – CVD experimental conditions used to deposit CrO_2 thin films

<i>Experimental parameters</i>	<i>Values</i>
Substrate temperature, T_s [$^\circ\text{C}$]	383 - 400
Precursor temperature, T_p [$^\circ\text{C}$]	260 - 295
Oxygen flow rate, ϕ_{O_2} [sccm]	180 - 500
Deposition time, t_{dep} [h]	1 - 6

The morphology and microstructure of the films were analysed by scanning electron microscopy (SEM). Their structure was studied by X-ray diffraction (XRD) with $\text{Cu K}\alpha$ radiation and by micro-Raman spectroscopy using the 514.5 nm excitation line of an Ar^+ laser. The magnetic characterization was carried out using a commercial SQUID magnetometer. Field cooling curves ($H=50\text{ Oe}$) for increasing temperatures between 35-400 K and hysteresis loops both for parallel and perpendicular applied magnetic fields at 35 K, were obtained for representative samples.

Results and discussion

Surface morphology and microstructure. The deposited films present good adherence, are homogeneous and exhibit a black shiny metallic colour. SEM analyses of different films have shown that their morphology consist of a uniform and very fine grain structure only resolved at high magnification (Fig. 1). For films grown with $T_s=390\text{ }^\circ\text{C}$, $T_p=260\text{ }^\circ\text{C}$, $\phi_{\text{O}_2}=280\text{ sccm}$ and $t_{\text{dep}}=6\text{ h}$ a uniform thickness of $260\pm 10\text{ nm}$ was measured, yielding a deposition rate of $0.72\pm 0.03\text{ nm}\cdot\text{min}^{-1}$.

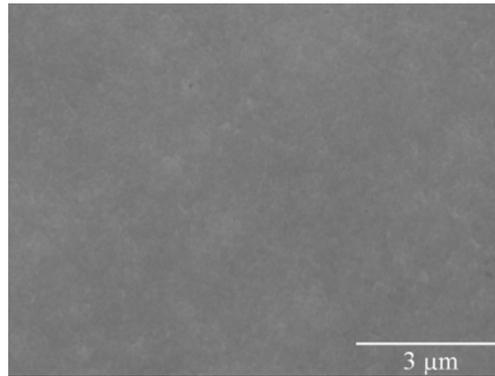


Fig. 1 – Scanning electron micrograph of a film deposited with $T_s=390\text{ }^\circ\text{C}$, $T_p=260\text{ }^\circ\text{C}$, $\phi_{\text{O}_2}=280\text{ sccm}$ and $t_{\text{dep}}=6\text{ h}$. Magnification: $\times 8000$

Structural analysis. Fig. 2 displays the XRD pattern of a film grown during 6 hours. Concerning CrO_2 deposition, it can be seen that only the (200) and (400) diffraction peaks of CrO_2 are clearly identified, revealing highly a-axis oriented film growth. The diffraction pattern also shows the presence of peaks from the Cr_2O_3 phase, (006) and (0012), although less intense than those of CrO_2 , indicating that the chromia phase is also highly oriented.

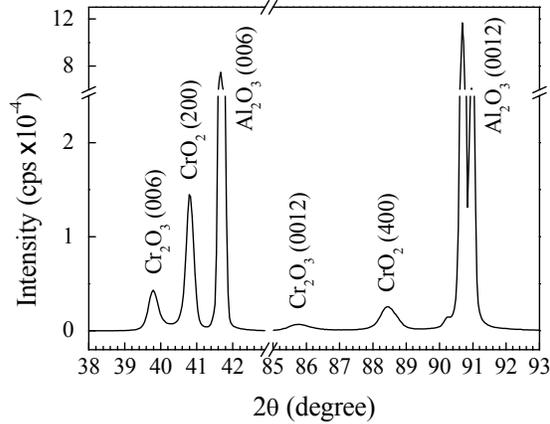


Fig. 2 – X-ray diffractogram of a CrO₂ film deposited with $T_s=390\text{ }^\circ\text{C}$, $T_p=260\text{ }^\circ\text{C}$, $\phi_{O_2}=280\text{ sccm}$ and $t_{dep}=6\text{ h}$.

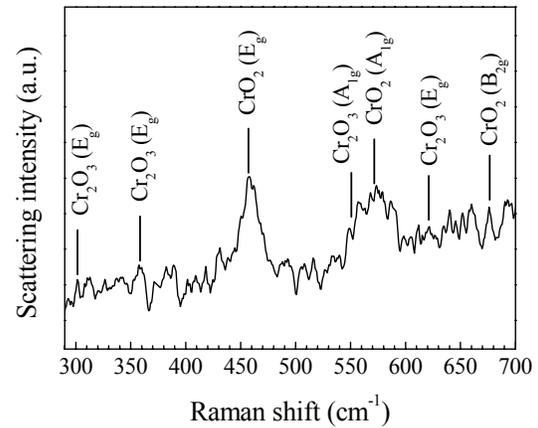


Fig. 3 – Micro-Raman spectrum of the film shown in Fig. 2.

In order to understand the location of the Cr₂O₃ phase in the sample, Raman microprobe studies were carried out by recording Raman spectra over different zones on the film surface. Fig. 3 shows a typical micro-Raman spectrum where two characteristic bands of the chromium dioxide are clearly observed at 458 cm^{-1} (E_g mode) and 573 cm^{-1} (A_{1g} mode) Raman shifts [13].

If some Cr₂O₃ were on the film surface it would be easily detected at the Raman shifts labeled on the figure, as the Raman scattering cross-section of Cr₂O₃ is much larger than that of CrO₂. Since this is not the case for the spectrum in Fig. 3, we can speculate that the Cr₂O₃ forms a layer at the CrO₂ film/substrate interface. A set of experiments was carried out varying the deposition time, in order to study the evolution of both chromium oxide phases as the films grow.

Fig. 4 presents X-ray diffractograms of films grown for different deposition times. After one hour of deposition only the Cr₂O₃ phase is observed via the (006) and (0012) diffraction lines. As deposition time increases, the intensities of the (200) and (400) peaks of the CrO₂ phase continuously increase in relation to those of Cr₂O₃ that stabilise after 2 hours. This result is consistent with Cr₂O₃ being formed as an interface layer between CrO₂ and the sapphire substrate.

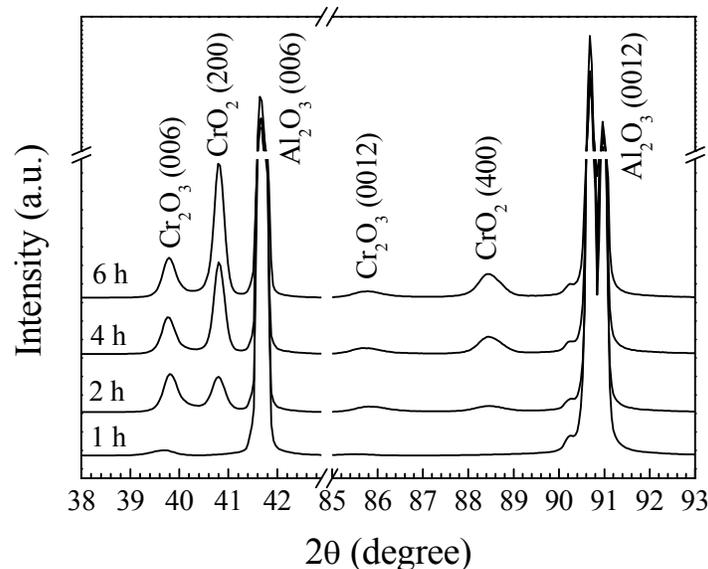


Fig. 4 – X-ray diffractograms of films deposited with $T_s=390\text{ }^\circ\text{C}$, $T_p=260\text{ }^\circ\text{C}$, $\phi_{O_2}=280\text{ sccm}$ and deposition times ranging from 1 to 6 h.

The preferred growth of Cr₂O₃ oxide phase in the early stages of the deposition process was previously reported by M. Rabe et al. [11]. These authors considered that Cr₂O₃, which is the stable

phase in the Cr-O phase diagram at atmospheric pressure, develops between the onset of the precursor temperature at $\sim 260^\circ\text{C}$ and the instant at which the substrate attains the required temperature of $\sim 390^\circ\text{C}$ for film growth. By contrast, in our experiments the deposition temperature is always attained before the CrO_3 melts at $\sim 196^\circ\text{C}$. Therefore, the growth of the Cr_2O_3 layer onto the sapphire substrate cannot be attributed to inadequate temperature differences between the precursor and the growing film. Instead, it can be explained by the close structural match between Cr_2O_3 and the Al_2O_3 substrate. Both compounds have hexagonal structure and close interatomic distances in the (001) plane: $a_{\text{Cr}_2\text{O}_3} = 0.4959\text{ nm}$ and $a_{\text{Al}_2\text{O}_3} = 0.4759\text{ nm}$, leading to a lattice mismatch of only 4.2%. Therefore, the chromia layer is slightly compressed in the film plane and expanded along the c-axis: the c parameter calculated from the diffractogram of the film grown during the 1st hour (Fig. 4) is 0.15% larger than the expected bulk value.

However, this small mismatch value is responsible by an increase of the layer energy as elastic strain energy. As the film thickness increases, the accumulated energy increases up to a critical thickness value for which the excess energy is sufficient to induce transition to a new phase, i.e. CrO_2 . It is interesting to note that the distance between the atomic positions with coordinates $10\bar{1}0$ and $\bar{1}\bar{1}20$ (0.8589 nm) differs of only 1.8% from the triple of the c parameter of CrO_2 lattice. This may justify an orientation relationship of the type $(100)_{\text{CrO}_2} \parallel (001)_{\text{Cr}_2\text{O}_3}$ and $[001]_{\text{CrO}_2} \parallel [210]_{\text{Cr}_2\text{O}_3}$, thus explaining the XRD results. Therefore, the growth of a Cr_2O_3 buffer layer between the CrO_2 and the sapphire substrate seems to be an intrinsic feature of the chosen film/substrate system.

Magnetic characterisation. Despite the presence of the Cr_2O_3 buffer layer, the CrO_2 films display a strong ferromagnetic behaviour. Fig. 5 shows the magnetization curve as a function of temperature for the film deposited during 6 h. As can be seen, a sharp magnetic transition is observed for $T \approx 375\text{ K}$ as determined from the inflection point of the magnetization versus temperature curve^a. Fig. 6 shows the magnetic hysteresis loop of the same film, measured at 35 K with the applied magnetic field parallel to the substrate surface. From this curve a coercive field of 110 Oe and a saturation magnetization of $624\text{ emu}\cdot\text{cm}^{-3}$ were determined, the later corresponding to $1.92\ \mu_{\text{B}}/\text{f.u.}$ close to the bulk value of $2\ \mu_{\text{B}}/\text{f.u.}$ for the CrO_2 . The analysis of hysteresis loops recorded both for parallel and perpendicular applied magnetic fields showed that the magnetization easy-axis is in the film plane.

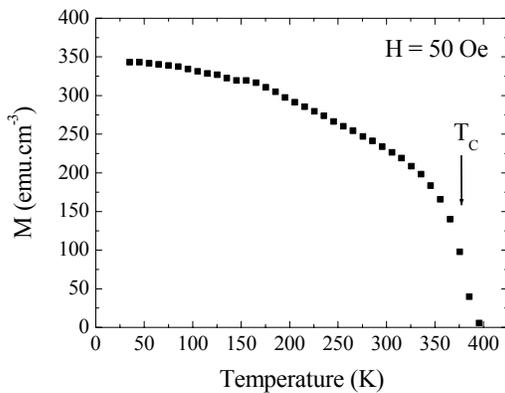


Fig. 5 – Magnetization vs. temperature for a CrO_2 film deposited with $T_s=390^\circ\text{C}$, $T_p=260^\circ\text{C}$, $\phi_{\text{O}_2}=280\text{ sccm}$ and $t_{\text{dep}}=6\text{ h}$, measured at $H=50\text{ Oe}$.

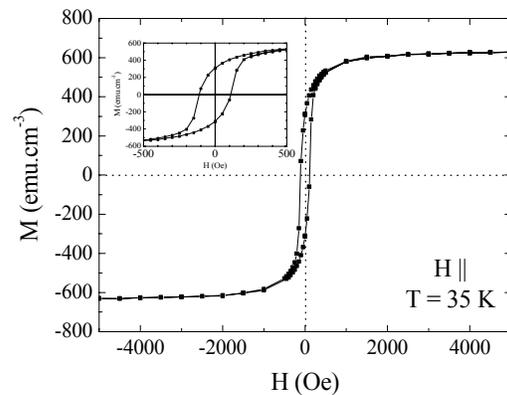


Fig. 6 – Magnetic hysteresis curve of a CrO_2 film deposited with $T_s=390^\circ\text{C}$, $T_p=260^\circ\text{C}$, $\phi_{\text{O}_2}=280\text{ sccm}$ and $t_{\text{dep}}=6\text{ h}$, measured at $T=35\text{ K}$ with the field parallel to the substrate surface. The inset shows the low field hysteresis loop for the same film.

Conclusions

Highly oriented CrO_2 thin films were deposited onto $\text{Al}_2\text{O}_3(0001)$ substrates by atmospheric

^a If T_c is taken as the temperature for which $M \rightarrow 0$ [10], the value deduced from Fig. 5 is $\sim 396\text{ K}$.

pressure CVD, using CrO₃ and O₂. The growth of Cr₂O₃ oxide as a buffer layer at the CrO₂/sapphire interface was shown and explained on the basis of structural match between the oxide and the substrate. Despite the existence of a Cr₂O₃ interlayer, the films display a sharp magnetic transition at 375 K and a saturation magnetization of 1.92 μ_B/f.u..

Acknowledgements

This work is supported by FCT (Portugal) contract POCTI/CTM/41413/2001 and EU contract FENIKS: G5RD-CT-2001-00535.

References

- [1] K.J. Schwartz, J. Phys. F: Met. Phys. Vol. 16 (1986), p. L211.
- [2] M.A. Korotin, V.I. Anisimov, D.I. Khomskii, and G.A. Sawatzky, Phys. Rev. Lett. Vol. 80 (1998) p. 4305.
- [3] R.J. Soulen, Jr., J.M. Byers, M.S. Osofsky, B.Nadgorny, T. Ambrose, S.F. Cheng, P.R. Broussard, C.T. Tanaka, J. Nowak, J.S. Moodera, A. Barry, and J.M.D. Coey, Science Vol. 282 (1998) p. 85.
- [4] Y. Ji, G.J. Strijkers, F.Y. Yang, C.L. Chien, J.M. Byers, A. Anguelouch, G. Xiao, and A. Gupta, Phys. Rev. Lett. Vol. 86 (2001) p. 5585.
- [5] A. Gupta and J.Z. Sun, J. Magn. Magn. Mater Vol. 200 (1999) p. 24.
- [6] S. Ishibashi, T. Namikawa and M. Satou, Mat. Res. Bull. Vol. 14 (1979) p. 51
- [7] Y.S. Dedkov, M. Fonine, C. König, U. Rüdiger, G. Güntherodt, S. Senz and D. Hess, Appl. Phys. Lett. Vol. 80 (2002) p. 4181.
- [8] A. Anguelouch, A. Gupta and G. Xiao, IEEE Trans. Magn. Vol. 37 (2001) p. 2135.
- [9] F.Y. Yang, C.L. Chien, X.W. Li, G. Xiao and A. Gupta, Phys. Rev. B Vol. 63 (2001) p. 92403.
- [10] H.A. Bullen and S.J. Garrett, Chem. Mater Vol. 14 (2002) p. 243.
- [11] M. Rabe, J. Dreßen, D Dahmen, J. Pommer, H. Stahl, U. Rüdiger, G. Güntherodt, S. Sens, D. Hess, J. Magn. Magn. Mater. Vol. 211 (2004) p. 314.
- [12] P.G. Ivanov, Ph.D. thesis, The Florida State University, 2002, p. 65.
- [13] M.N. Iliev, A.P. Litvinchuk, H.G. Lee, C.W. Chu, A. Barry and J.M.D. Coey, Phys. Stat. Sol. (b) 215 (1999) 643.